

Increase in atmospheric CO₂ has caused an increase in ocean acidification

GOA-ON Goal 2: Improve our understanding of ecosystems response to ocean acidification

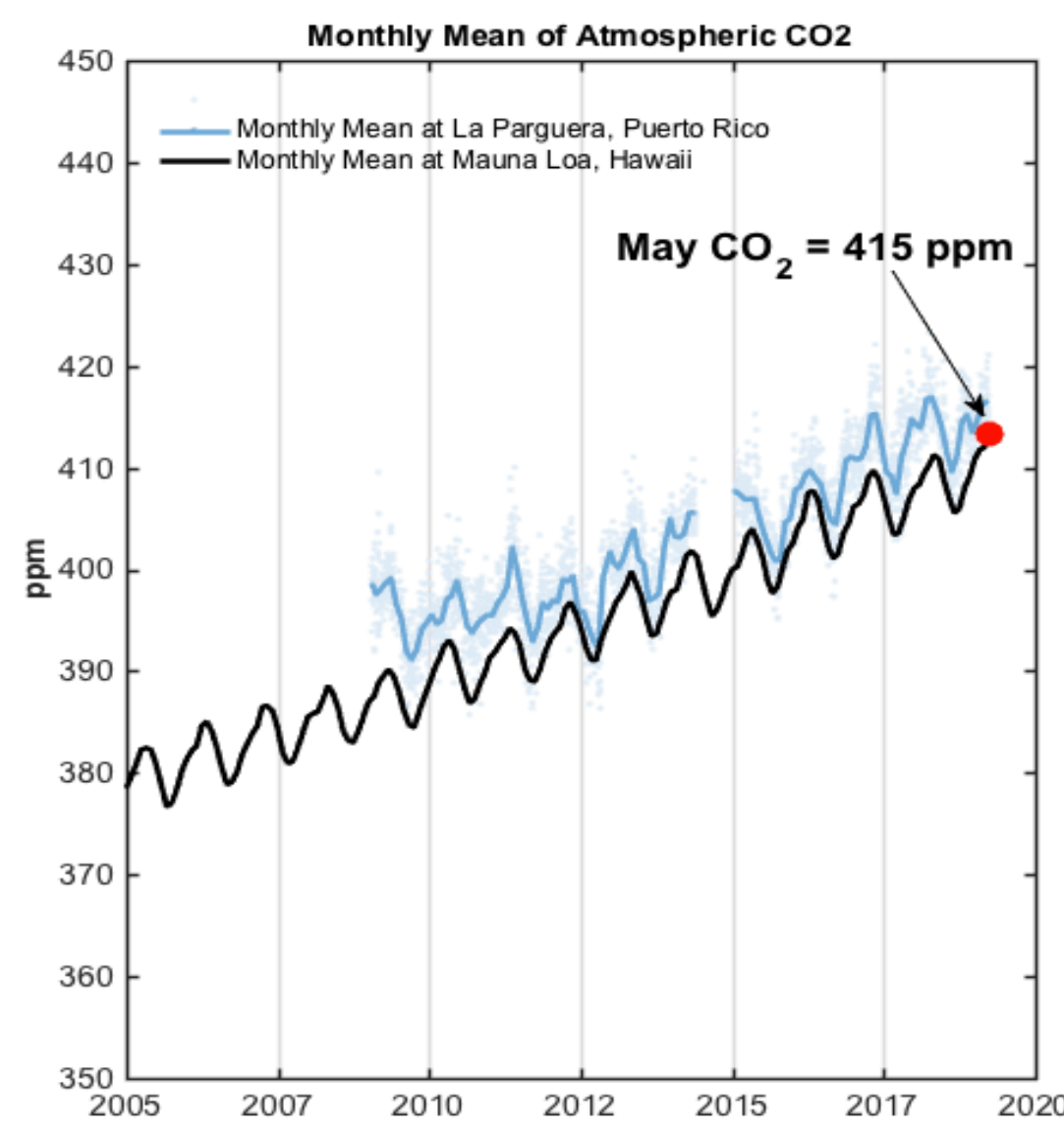


Fig. 1: CO₂ concentration in Air

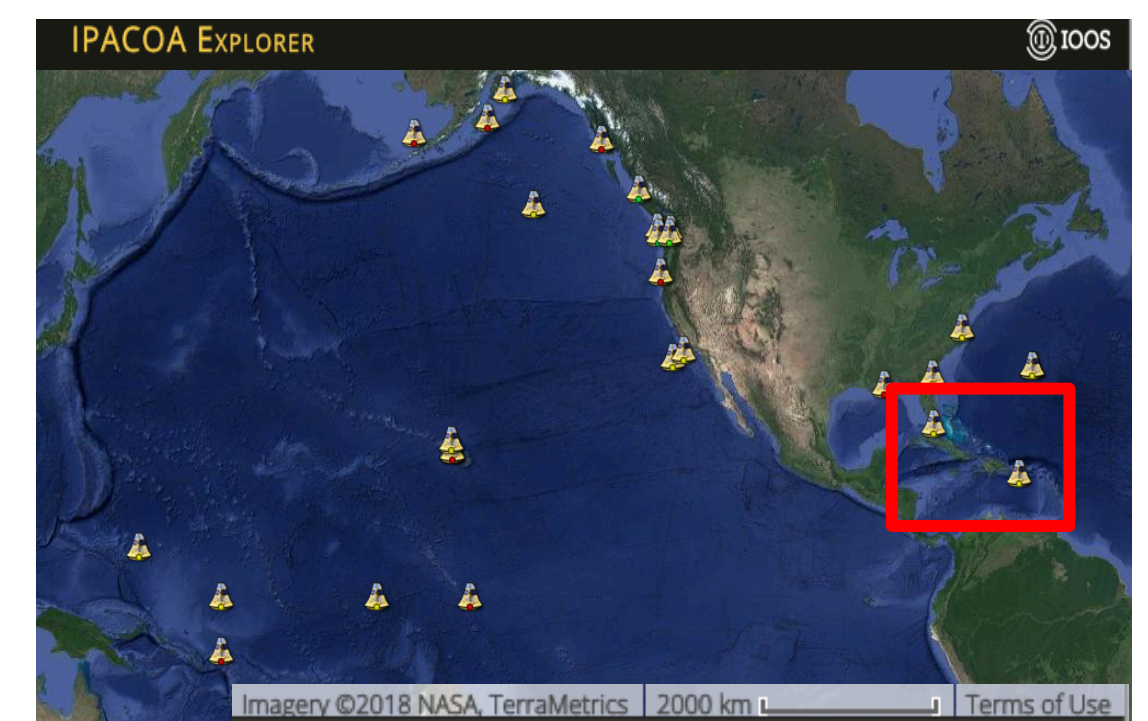
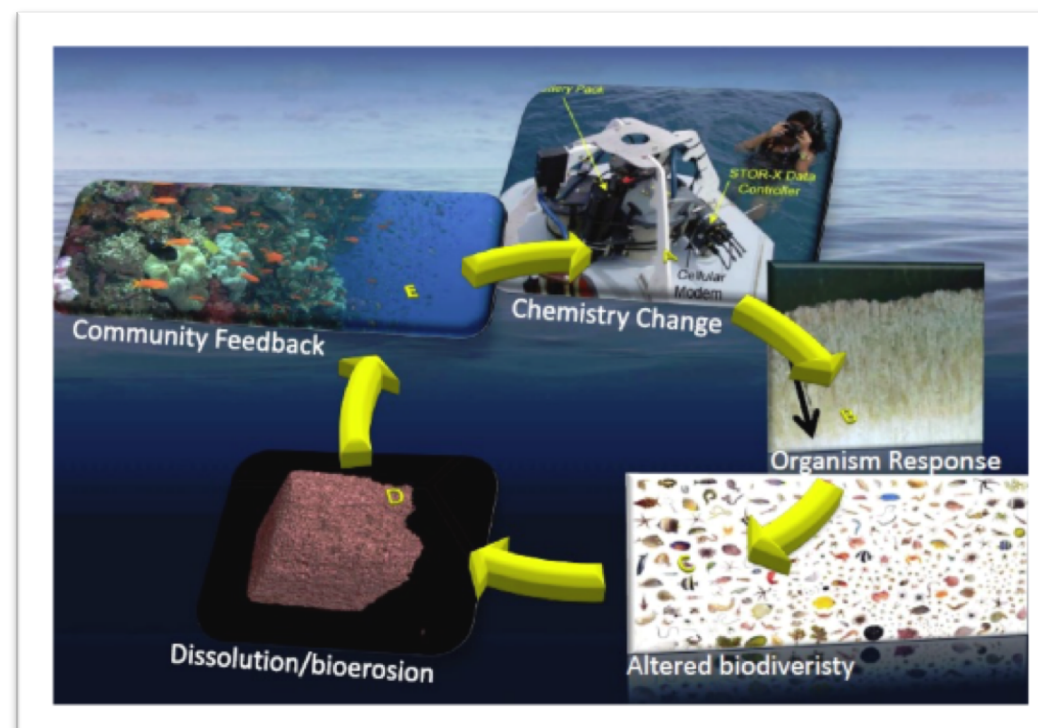


Fig. 2: Global Ocean Acidification Observing Network – GOA-ON Caribbean Region include two buoys located in Puerto Rico & Florida

Monitoring the Chemical Change in Coral Reefs



Fig.3: MapCO2 buoy

The MapCO₂ buoys provides measurements of CO₂ in the air and seawater, pH, temperature, and salinity every 3 hours.

A primary concern with respect to coral reef ecosystems is coral health and their ability to precipitate calcium carbonate.

Community & Environmental Feedbacks

Metabolic Rates

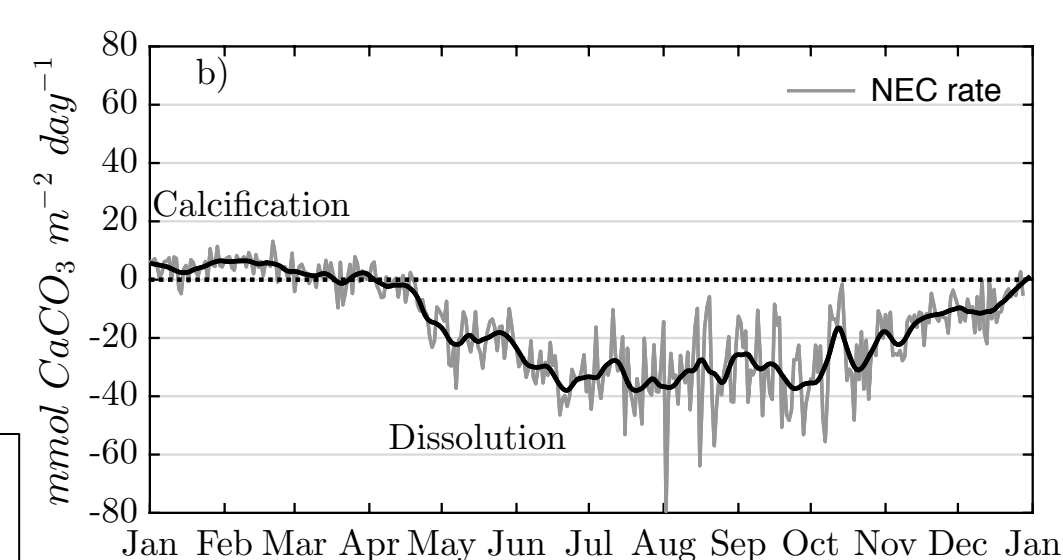
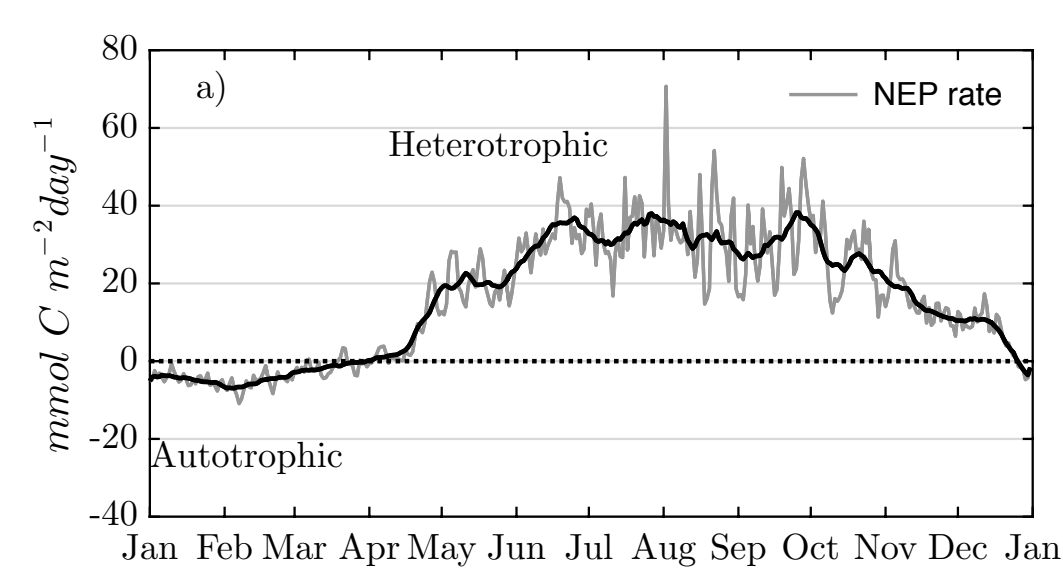
Fig.4: Biogeochemical processes affecting the carbonate chemistry

	Alk	DIC	pCO ₂	pH	[HCO ₃ ⁻]	[CO ₃ ²⁻]	comments
increase in atmospheric CO ₂	no change	↑	↑	↓	↑	↓	causes CaCO ₃ dissolution
photosynthesis	(+)	↓	↓	↑	↓	↑	nitrate uptake slight effect on alkalinity
respiration	(-)	↑	↑	↓	↑	↓	nitrate release slight effect on alkalinity
calcification	↓	↓	↑	↓	↓	↓	usually coupled with photosynth or respiration
CaCO ₃ dissolution	↑	↑	↓	↑	↑	↑	affected by respiration

↑ primary changes ↓ responses (+), (-) minor responses

EPOC/BIOACID/CaMar/O'CB training workshop - Best practices in ocean acidification research

Puerto Rico – La Parguera



Florida – Cheeca Rocks

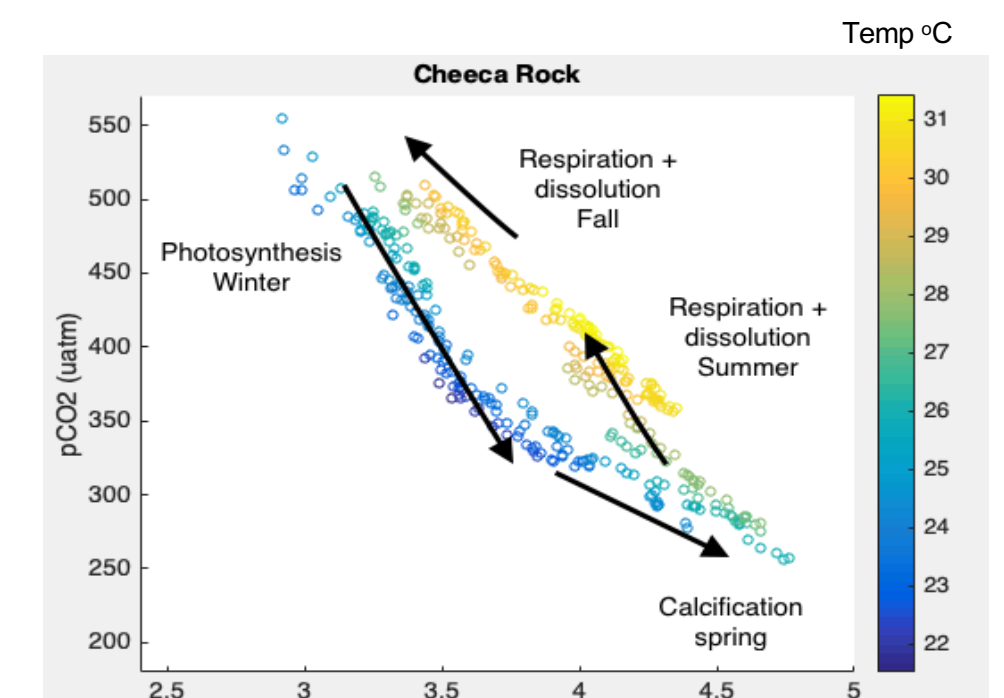
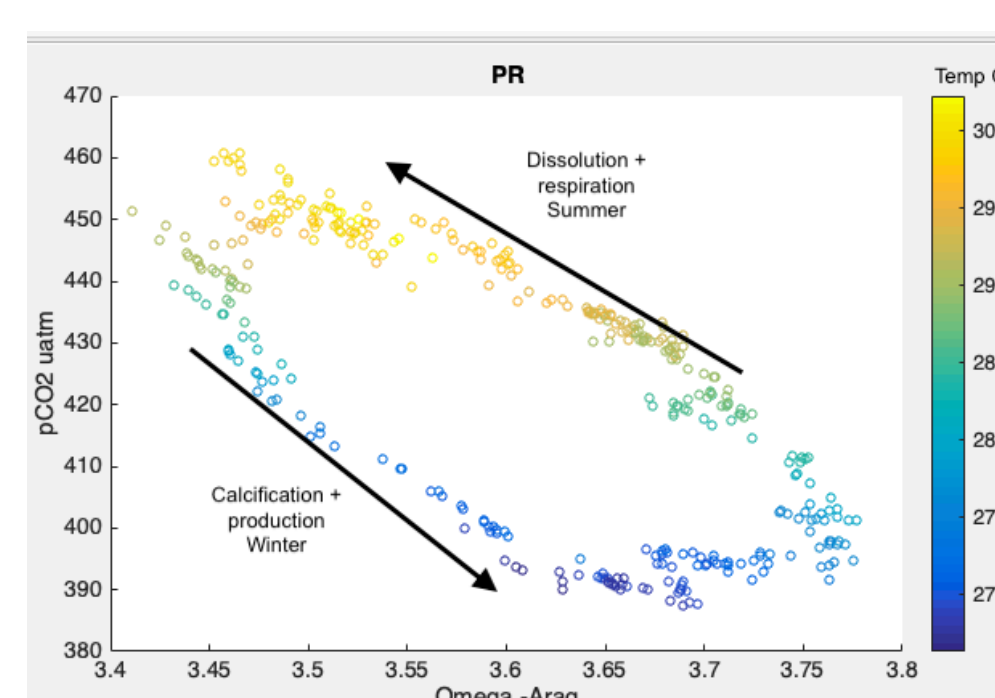
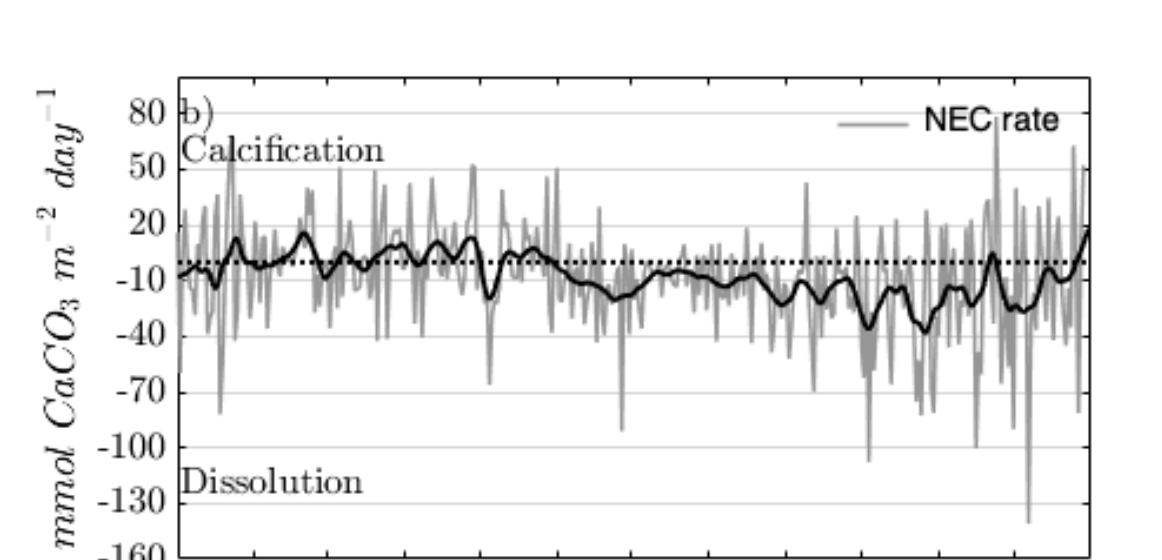
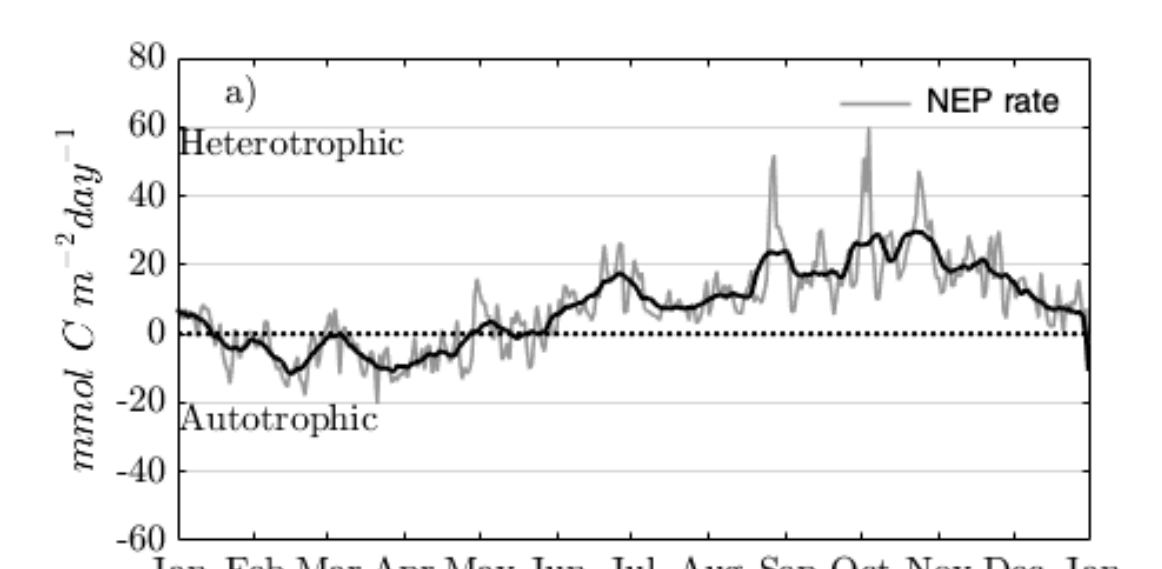


Fig.5: Annual composites based on daily averages of modeled a) NEP (mmol C m⁻² day⁻¹; gray line) and b) NEC rates (mmol CaCO₃ m⁻² day⁻¹; gray line). Seawater pCO₂ vs Ω_{arag} and seawater temperature (color bar) for La Parguera, PR and the Cheeca Rocks, Florida.

Remarks:

- Tropical Caribbean reef ecosystems are likely exhibiting periods of net dissolution during the summer time.
- The dissolution and heterotrophic events overlapped with the beginning of the shellfish spawning period.
- The annual carbonate dissolution rate in both studied sites ranged between -0.5 to -0.62 kg CaCO₃ m⁻² yr⁻¹.
- The increase of temperature increases the respiration and dissolution rates and the pCO₂ over the summer months.
- Autonomous capabilities of buoyed operational systems such as these can be used to detect metabolic processes sensitive to ocean acidification.